

KEY - CHEMISTRY pH PRACTICE (#2)

- 1) How acidic or basic a substance is... (answers vary)
 - 2) 14
 - 3) 0
 - 4) 7
 - 5) A pH of 6.7 is a weak acid. It is less than 7, but barely.
 - 6) A pH of 13.1 a strong base. It is almost 14.
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- 7) A solution has $[\text{OH}^-]$ of 1.00×10^{-9} M.
 - a) $[\text{H}^+] = 1.00 \times 10^{-5}$ M
 - b) $\text{pH} = 5$
 - c) $\text{pOH} = 9$
 - d) ACID ($\text{pH} = 5$)
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- 8) A solution has $[\text{H}^+] = 1.00 \times 10^{-11}$ M.
 - a) $[\text{OH}^-] = 1.00 \times 10^{-3}$ M
 - b) $\text{pOH} = 3$
 - c) $\text{pH} = 11$
 - d) BASE ($\text{pH} = 11$)
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- 9) A solution has a pOH of 8.
 - a) $\text{pH} = 6$
 - b) $[\text{H}^+] = 1.0 \times 10^{-6}$ M
 - c) $[\text{OH}^-] = 1.0 \times 10^{-8}$ M
 - d) ACID ($\text{pH} = 6$)
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- 10) A solution has a $\text{pH} = 7$.
 - a) $[\text{OH}^-] = 1.0 \times 10^{-7}$ M
 - b) $\text{pOH} = 7$
 - c) $[\text{H}^+] = 1.0 \times 10^{-7}$ M
 - d) NEUTRAL ($\text{pH} = 7$)
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(#11-20 for Chem IH only)

- 11) A solution has $[\text{H}^+] = 3.39 \times 10^{-7}$ M.
 - a) $[\text{OH}^-] = 2.95 \times 10^{-8}$ M
 - b) $\text{pOH} = 7.53$
 - c) $\text{pH} = 6.47$
 - d) ACID ($\text{pH} < 7$)
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12) A solution has a pOH of 8.55.

- a) pH = 5.45
 - b) $[H^+] = 3.55 \times 10^{-6} M$
 - c) $[OH^-] = 2.82 \times 10^{-9} M$
 - d) ACID (pH < 7)
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13) A solution has $[OH^-]$ of $4.44 \times 10^{-6} M$.

- a) $[H^+] = 2.24 \times 10^{-9} M$
 - b) pH = 8.65
 - c) pOH = 5.35
 - d) BASE (pH > 7)
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14) A solution has a pH = 12.76.

- a) pOH = 1.24
 - b) $[H^+] = 1.74 \times 10^{-13} M$
 - c) $[OH^-] = 5.75 \times 10^{-2} M$
 - d) BASE (pH > 7)
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15) A solution has a pOH of 10.75.

- a) pH = 3.25
 - b) $[H^+] = 5.62 \times 10^{-4} M$
 - c) $[OH^-] = 1.78 \times 10^{-11} M$
 - d) ACID (pH < 7)
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16) A solution has $[H^+] = 5.12 \times 10^{-8} M$.

- a) $[OH^-] = 1.95 \times 10^{-7} M$
 - b) pOH = 6.71
 - c) pH = 7.29
 - d) BASE (pH > 7)
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17) A pH change from 8 \rightarrow 12 is 4 increments on the scale.

$10^4 = 10,000$ times more basic

18) A pH change from 5 \rightarrow 3 is 2 increments on the scale.

$10^2 = 100$ times more acidic

19) A pH change from 9 \rightarrow 6 is 3 increments on the scale.

$10^3 = 1000$ times less basic

20) A pH change from 0 \rightarrow 1 is 1 increment on the scale.

$10^1 = 10$ times less acidic